Research paper

Corundum formation by metasomatic reactions in Archean metapelite, SW Greenland: Exploration vectors for ruby deposits within high-grade greenstone belts

Chris Yakymchuk a, Kristoffer Szilas b, *

a Earth and Environmental Sciences, University of Waterloo, 200 University Ave West, Waterloo, Ontario N2L 3G1, Canada
b Geological Survey of Denmark and Greenland (GEUS), Øster Voldgade 10, 1350 Copenhagen K, Denmark

ARTICLE INFO

Article history:
Received 27 March 2017
Received in revised form 29 June 2017
Accepted 29 July 2017
Available online xxx

Keywords:
Ultramafic rocks
Kyanite
Sillimanite
Metasomatism
Pseudosection
Desilification

ABSTRACT

Corundum (ruby-sapphire) is known to have formed in situ within Archean metamorphic rocks at several localities in the North Atlantic Craton of Greenland. Here we present two case studies for such occurrences: (1) Maniitsoq region (Kangerdluarssuk), where kyanite paragneiss hosts ruby corundum, and (2) Nuuk region (Store), where sillimanite gneiss hosts ruby corundum. At both occurrences, ultramafic rocks (amphibole-peridotite) are in direct contact with the ruby-bearing zones, which have been transformed to mica schist by metasomatic reactions. The bulk-rock geochemistry of the ruby-bearing rocks is consistent with significant depletion of SiO₂ in combination with addition of Al₂O₃, MgO, K₂O, Th and Sr relative to an assumed aluminous precursor metapelite. Phase equilibria modelling supports ruby genesis from the breakdown of sillimanite and kyanite at elevated temperatures due to the removal of SiO₂. The juxtaposition of relatively silica- and aluminum-rich metasedimentary rocks with low silica ultramafic rocks established a chemical potential gradient that leached/mobilized SiO₂ allowing corundum to stabilize in the former rocks. Furthermore, addition of Al₂O₃ via a metasomatic reaction is required, because Al/Ti is fractionated between the aluminous precursor metapelites and the resulting corundum-bearing mica schist. We propose that Al was mobilized either by complexation with hydroxide at alkaline conditions, or that Al was transported as K-Al-Si-O polymers at deep crustal levels. The three main exploration vectors for corundum within Archean greenstone belts are: (1) amphibolite-to-granulite-facies metamorphic conditions, (2) the juxtaposition of ultramafic rocks and aluminous metapelites, and (3) mica-rich reactions zones at their interface.

© 2017, China University of Geosciences (Beijing) and Peking University. Production and hosting by Elsevier B.V. This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/4.0/).

1. Introduction

Corundum (Al₂O₃) is a relatively rare metamorphic mineral that requires unusual geochemical conditions of low silica activity combined with high aluminum contents of the host rock. Corundum in its purest form is colorless, but when trace element impurities of Cr, Fe or V enter the structure, corundum can become the colored gem ruby (red) or sapphire (blue), although other colors are also possible (e.g., Simonet et al., 2008). High quality gem ruby can be as expensive as diamond (e.g., Shor and Weldon, 2009);

hence it is of economic interest for mineral exploration companies, as well as for small-scale miners.

A major concern for gemologists is to identify the origins of gem quality corundum in order to control the trade of ruby and sapphire to avoid illegal activities. In recent years, several analytical methods have been employed with the aim of fingerprinting the origins of cut gem stones, including oxygen isotope analysis, Raman spectroscopic analysis (for inclusions), and in situ trace element analysis (e.g., Porto and Krishnan, 1967; Giuliani et al., 2005, 2014; Pornwilard et al., 2011). It has been demonstrated that corundum from Greenland has a rather unique composition in a global context, including mantle-like O-isotope compositions and elevated Cr and Si, which allows for precise fingerprinting of cut gems originating from the North Atlantic Craton (Thirangoon, 2008; Keulen and Kalvig, 2013). This specific signature reflects
the unusual geologic conditions in which Greenlandic ruby has formed, as we will give examples of in this paper.

One of the fundamental issues relevant for corundum prospecting is to identify the combination of rock types and metamorphic environment that favors its formation. Many of the most economic ruby occurrences are placer deposits sourced from marbles hosting corundum or alkaline dykes containing corundum xenocrysts (e.g., Sutherland et al., 1998; Garnier et al., 2008; Groat et al., 2014). In metamorphic terranes, proposed mechanisms for corundum stabilization include hydrothermal alteration (e.g., Bottrill, 1998), metasomatic exchange of silica with the ultramafic rocks (e.g., Riesco et al., 2005), and anatexis of aluminous protoliths accompanied by melt loss (e.g., Cartwright and Barnicoat, 1986; Palke et al., 2017).

In the present study, we highlight a less common primary geological environment of corundum formation, namely high-grade Archean greenstone belts. We present two case studies from southern West Greenland, where ruby corundum is hosted by aluminous Archean metapelite. Thermodynamic modelling is used to show under which pressure, temperature and geochemical conditions such rocks can stabilize corundum, and we demonstrate the critical importance of associated ultramafic rocks, which act as chemical buffers to maintain a low silica activity in the system. We summarize our findings by proposing several exploration vectors for corundum deposits within Archean greenstone belts, which may have applications for similar geological environments globally.

2. Geologic setting

The main portion of the North Atlantic Craton occurs along the SW coast of Greenland (Fig. 1). It is dominated by high-grade grey orthogneiss of the tonalite-trondhjemite-granodiorite (TTG) suite, although greenstone belts and fragmented anorthosite complexes also comprise an important component (e.g., Windley and Garde, 2009). Given that the Archean greenstone belts in this region have all experienced amphibolite- to granulite-facies metamorphism (e.g., McGregor and Friend, 1992; Garde, 1997), the neutral term ‘supracrustal belt’ has traditionally been preferred for such rock associations, and we therefore use this ‘local’ nomenclature here.

The geodynamic environment of formation for Archean crust is debated and still controversial within the geologic community. Some researchers argue that uniformitarian principles can be applied back into the Eoarchean (e.g., Polat et al., 2002; Komiya et al., 2015) or even the Hadean (Harrison et al., 2005; Watson and Harrison, 2005; Hopkins et al., 2008), while others argue that modern-style plate tectonics did not operate until the Mesoarchean (Smithies et al., 2007; Dhuime et al., 2012) or perhaps even the Neoproterozoic (Brown, 2008; Stern, 2008; Hamilton, 2011). There are several Archean cratons which clearly did not form by uniformitarian processes, one such example being the East Pilbara Craton (Collins et al., 1998; Van Kranendonk et al., 2004). However, the North Atlantic Craton may represent the opposite case, in which even the oldest component potentially formed via geologic processes resembling those of modern-style subduction zone systems. The currently favored model for the formation of the Archean continental crust in Greenland is by subduction (e.g., Nutman et al., 1996, 2015; Garde, 1997), and it is also the preferred interpretation for the formation of supracrustal belts (e.g., Garde, 2007; Polat et al., 2008; Jenner et al., 2009; Szilas et al., 2012, 2013, 2017a), as well as for the associated anorthosite complexes of the North Atlantic Craton (Polat et al., 2009, 2012; Hoffmann et al., 2012; Huang et al., 2014). Although the question of the overall geodynamic environment is not the main topic of the present contribution, it does have some implications for the interpretation of the metamorphic evolution of the rocks describe and investigated in the present study.

Corundum is a high-grade metamorphic mineral that has been reported from several localities within Archean rocks in southern West Greenland. The best-known example is the extensive ruby

Figure 1. Geological map of the Maniitsoq and Nuuk regions of southern West Greenland. Based on mapping by the Geological Survey of Denmark and Greenland (GEUS).
deposits within the Fiskenæsset Anorthosite Complex (Myers, 1985; Polat et al., 2009), where corundum occurs along contacts between leucogabbro and ultramafic rocks and is interpreted to have formed during metasomatism at high-grade metamorphic conditions, although some occurrences also indicate the involvement of late tectonic granitic pegmatites (Keulen et al., 2014).

Minor corundum formation is also documented within the Eoarchean Itsaq Gneiss Complex (Appel et al., 2002; Lowry et al., 2003). In this case oxygen isotopes indicate an external fluid input to the system, where late metamorphic ruby corundum formed at the interface between amphibole-harzburgite and intrusive granitoid sheets.

Less well-known corundum occurrences are present on the island of Storø in the Nuuk region (Van Gool, 2006), where the mineral occurs within mica schist at the contact between metapelites and ultramafic rocks (Fig. 2). The Storø Supracrustal Belt (SSB) has received much attention due to the presence of gold mineralization hosted by sheeted quartz veins within amphibolites (see a summary in Scherstén et al., 2012). The maximum age of the SSB is constrained at ca. 2800 Ma by U-Pb dating of detrital zircon in quartzite (Scherstén et al., 2012; Szilas et al., 2014b). Re-Os isotope data on arsenopyrite demonstrates that the gold precipitated in quartz veins during regional metamorphism, and likely formed at two distinct events at ca. 2707 Ma and ca. 2640 Ma, respectively (Scherstén et al., 2012). The aluminous garnet-sillimanite-biotite gneiss that appears to be the precursor for the corundum-bearing rocks on Storø (see Section 3) has been investigated intensely, because it was debated if this rock type represents a hydrothermal alteration product or simply a metapelite with a mafic source (e.g., Knudsen et al., 2007; Szilas and Garde, 2013; Szilas et al., 2016). Corundum formation in these rocks likely took place during peak metamorphic conditions at ca. 2635 Ma, given by U-Pb ages of zircon rims found consistently in all felsic rocks of the SSB (Hollis, 2005; Nutman et al., 2007; Van Gool et al., 2007). The Storø corundum occurrence is the first case study treated in the present work.

The second case study that we present in this contribution concerns a corundum occurrence in the Maniitsoq region (Garde and Marker, 1988), which is also hosted by metapelite juxtaposed with ultramafic rocks, which are high-grade amphibole-peridotites (Fig. 3). The Maniitsoq region was first studied extensively by Ramberg (1948), Sørensen (1954) and Berthelsen (1962), who documented several occurrences of ultramafic enclaves within orthogneiss as well as the presence of unusual metamorphic minerals such as corundum and sapphirine.

The tonalitic orthogneisses of the Maniitsoq and the Akia terrane have igneous ages of around 3 Ga (Garde et al., 2000; Windley and Garde, 2009). Dyck et al. (2015) presented a model...
for the Maniitsoq region where the area hosting corundum forms part of the so-called Majorqaq Belt, which was interpreted to represent lower crustal remnants of a Neoarchean orogen. Migmatitic metapelites from the Majorqaq Belt experienced a clockwise P-T evolution with peak metamorphic conditions of >810 °C at ~9 kbar (Dyck et al., 2015). U-Pb ages of ~2.56 Ga from anatectic zircon were interpreted by Dyck et al. (2015) to reflect the minimum age of the accretion of the Majorqaq Belt to the proto-North Atlantic Craton.

A controversial claim of a giant meteorite impact structure hosted by the Maniitsoq region has been proposed by Garde et al. (2014). However, several rebuttals to that claim have been put forth (e.g., Reimold et al., 2013, 2014), and given the small-scale focus of the present study, we will not discuss the meteorite impact model further in this work.

3. Field observations and petrography

3.1. Størø field observations

The island of Størø is located in the Nuuk region (Fig. 1) and it hosts the Størø Supracrustal Belt (SSB), which represents our first case study. Fig. 2 shows a detailed geologic map of the corundum occurrence within the SSB. The proximity of ultramafic rocks with aluminous rocks containing abundant garnet and sillimanite should be noted as a prospective setting for corundum formation.

A distinct type of schist is the host of the corundum and it occupies a less than 2-m-wide zone between the ultramafic rock and the aluminous gneiss. Fig. 4 shows examples of the three main lithological units in the SSB, namely amphibolite, biotite schist and aluminous gneiss as also seen on the detailed map in Fig. 2. Panels b, d and f in Fig. 4 show the corundum-bearing mica schist found at the contact between the aluminous gneiss and the ultramafic rock of Fig. 2. The narrow mica schist that hosts corundum is generally less than 2 m wide and is thus not mappable even at this scale. Garnet is present within this rock on the side that is in contact with the aluminous gneiss, whereas corundum is only present on the side of the mica schist that is in contact with the ultramafic rock. It should be noted that although the mica schist is present along the entire contact between the ultramafic rock and the aluminous gneiss, it is generally only a thin (<1 m) reaction zone with no corundum. Only where the mica schist forms a thicker unit of up to 3 m, does it contain corundum.

3.2. Maniitsoq field observations

The corundum-bearing Kangerdluarssuk locality in the Maniitsoq region (Fig. 1) has previously been described by Garde and Marker (1988) and Dyck et al. (2015). For simplicity, we refer to the Kangerdluarssuk locality, as the Maniitsoq ruby locality throughout this work to distinguish it from the Størø ruby locality. The detailed geologic map presented in Fig. 3 shows that corundum is only found along the contact between the ultramafic rocks and the kyanite-bearing metapelite. A few other occurrences are also reported by Garde and Marker (1988) at this locality, although they were not located during our own field work and thus not included in this map. Sapphirine has also been reported in the Maniitsoq region by Ramberg (1948) and Sørensen (1954). However, this mineral was not observed by us at any of the localities we visited during field work in the region.

At Maniitsoq we also sampled several amphibolites, for which we present geochemical data. These amphibolites are heavily modified containing calc-silicates and garnet-plagioclase veins and they are generally strongly sheared (Fig. 5e).

Fig. 5 shows examples of the three main lithological units at the Maniitsoq ruby locality (Kangerdluarssuk), namely amphibolite, kyanite gneiss and ultramafic rock as also seen on the detailed map in Fig. 3. Panels b, d and f in Fig. 5 show the corundum-mica schist found at the contact between the aluminous gneiss and the ultramafic rock of Fig. 3. Fig. 5c shows a close-up of the ultramafic rocks adjacent to the corundum-bearing schist at the Maniitsoq locality. Note the brown unaltered peridotite patches between massive
tremolite veins, which are oriented orthogonally to the fracture orientations. In Table 1 we present geochemical data for both the unaltered (sample 455) and the amphibole-rich peridotites (samples 466 and 467).

### 3.3. Petrographic observations

The petrography of representative examples of the main lithological units relevant for the corundum formation at the Maniitsoq ruby locality are presented in Fig. 6. The peridotite consists mostly of olivine, spinel and orthopyroxene (Fig. 6a) with essentially no hydrous phases present, except for minor amphibole. This rock type is found at several localities in the Maniitsoq region, and detailed studies of similar rocks at Fiskefjord demonstrates these rocks to be represent cumulates related to dismembered layered intrusions (e.g., Szilas et al., 2015, 2017b). The ultramafic rock or amphibole-rich peridotite s.l. is dominated by fine-grained tremolitic amphibole (Fig. 6b), and is found as an extensive network of fractures in proximity to the ruby mica schist.

Kyanite gneiss is comprised mainly of quartz, kyanite (Fig. 6c and d) and garnet with plagioclase, mica, amphibole and spinel as minor constituents. Centimeter-scale quartz-rich layers alternate with blueish fibrolitic kyanite-rich layers (Fig. 6c and d) and kyanite may comprise up to 40 vol.% of this rock type.

Ruby mica schist (Fig. 6e and f) consists mainly of biotite with corundum and plagioclase being the main other minerals in this rock type. Locally orthopyroxene, olivine and amphibole are also present. Corundum contains aligned inclusions of biotite, indicating a relict foliation that is at a high angle to the main foliation. Mica also wraps around the corundum porphyroblasts, which suggests some deformation after corundum growth.

### 4. Methods

#### 4.1. Bulk-rock geochemistry

Bulk-rock major and trace element data for the samples from the Maniitsoq region and the ruby mica schist from Storø were obtained from ALS Global laboratory in Ireland using their ME-MS61 procedure, which includes measurement by Inductively Coupled Plasma (ICP) methods. Below a brief summary of this method is given as it appears on ALS’s website (www.alsglobal.com).

A prepared sample (0.25 g) is digested with perchloric, nitric, hydrofluoric and hydrochloric acids. The residue is topped up with dilute hydrochloric acid and analyzed by inductively coupled plasma-atomic emission spectrometry (ICP-AES). Following this analysis, the results are reviewed for high concentrations of bismuth, mercury, molybdenum, silver and tungsten and diluted accordingly. Samples meeting this criterion are then analyzed by ICP-MS, and results are corrected for spectral interelement interferences. Four acid digestions are able to dissolve most minerals; however, although the term “near-total” is used, depending on the sample matrix, not all elements are quantitatively extracted.

Bulk-rock geochemical data for the rocks from the SSB are from Szilas et al. (2016). One exception is sample 11801, which is the corundum-mica schist from this locality, which was also analyzed...
at ALS Global. All the new geochemical data are presented in Table 1 below and selected plots are shown in Figs. 7–9.

4.2. Computational methods

Phase equilibria modeling is used to (1) assess the pressure (P) and temperature (T) conditions of metamorphism, (2) explore the role of silica depletion on the mineral assemblage for the assumed precursors to the ruby-bearing rocks, (3) quantify the amount of silica depletion required to stabilize corundum, and (4) determine the amount and composition of melt expected at various P-T conditions and the composition of the residuum after melt loss. Calculations were conducted using THERMOCALC v.3.40 (Powell and Holland, 1988) using the internally consistent dataset (ds62) of Holland and Powell (2011). For the aluminous gneiss and biotite gneiss, modelling was undertaken in the MnO-Na2O-CaO-K2O-FeO-MgO-Al2O3-SiO2-H2O-TiO2-Fe2O3 (MnNCKFMASHTO) chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b). An amphibolite composition was investigated in the NCKFMASHTO chemical system using the activity-composition relations in White et al. (2014a,b).

For the samples from Storø, this pressure was set at 7 kbar, given the minimum estimate of Persson (2007) by garnet-biotite-plagioclase thermobarometry of rocks in the SSB recording temperatures of 521–624 °C and pressures of 4.5–6.1 kbar, which likely represent retrograde overprinting (minimum conditions). For the samples from Maniitsoq, the solidus was saturated with H2O at 10 kbar, which is similar to the calculations by Dyck et al. (2015) for amphibole-kyanite-bearing metapelites at Maniitsoq. If the modelled heating path crossed the solidus at lower or higher pressures, the quantity of melt produced will be slightly overestimated and underestimated, respectively, and minor variation in the phase assemblages are expected.
Table 1: Bulk-rock geochemical data for the Maniitsoq rocks and one ultramafic ruby schist from Stora (sample 11801). Major elements are in wt.%, trace elements are in ppm. Blank values are below the limit of detection.

<table>
<thead>
<tr>
<th>Sample#</th>
<th>Rock type</th>
<th>Latitude (° N)</th>
<th>Longitude (° E)</th>
<th>SiO2</th>
<th>Al2O3</th>
<th>Fe2O3</th>
<th>CaO</th>
<th>MgO</th>
<th>Na2O</th>
<th>K2O</th>
<th>Cr2O3</th>
<th>TiO2</th>
<th>MnO</th>
<th>P2O5</th>
<th>LOI</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td>454</td>
<td>Amphibolite</td>
<td>65.5619</td>
<td>-52.4008</td>
<td>53.1</td>
<td>19.10</td>
<td>7.02</td>
<td>11.90</td>
<td>5.88</td>
<td>0.16</td>
<td>0.30</td>
<td>0.35</td>
<td>0.21</td>
<td>0.01</td>
<td>0.62</td>
<td>0.01</td>
<td></td>
</tr>
<tr>
<td>455</td>
<td>Peridotite</td>
<td>65.5686</td>
<td>-52.3906</td>
<td>40.7</td>
<td>19.42</td>
<td>9.38</td>
<td>1.05</td>
<td>4.44</td>
<td>0.96</td>
<td>0.81</td>
<td>0.35</td>
<td>0.28</td>
<td>0.01</td>
<td>0.02</td>
<td>0.03</td>
<td></td>
</tr>
<tr>
<td>456</td>
<td>Ruby mica schist</td>
<td>65.5638</td>
<td>-52.3927</td>
<td>34.4</td>
<td>33.2</td>
<td>10.82</td>
<td>1.05</td>
<td>13.05</td>
<td>0.94</td>
<td>1.17</td>
<td>0.11</td>
<td>0.08</td>
<td>0.01</td>
<td>0.01</td>
<td>0.01</td>
<td></td>
</tr>
<tr>
<td>457</td>
<td>Calc-silicate amphibolite</td>
<td>65.5673</td>
<td>-52.3925</td>
<td>48.4</td>
<td>63.2</td>
<td>11.70</td>
<td>5.66</td>
<td>7.14</td>
<td>0.68</td>
<td>0.24</td>
<td>0.17</td>
<td>0.21</td>
<td>0.01</td>
<td>0.04</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>458</td>
<td>Leuco-amphibolite</td>
<td>65.5673</td>
<td>-52.3921</td>
<td>51.0</td>
<td>10.20</td>
<td>10.20</td>
<td>5.50</td>
<td>5.21</td>
<td>0.28</td>
<td>0.05</td>
<td>0.16</td>
<td>0.12</td>
<td>0.03</td>
<td>0.15</td>
<td>0.21</td>
<td></td>
</tr>
<tr>
<td>459</td>
<td>Ultramafic rock</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>57.5</td>
<td>18.20</td>
<td>8.99</td>
<td>2.90</td>
<td>3.73</td>
<td>0.30</td>
<td>0.10</td>
<td>0.04</td>
<td>0.09</td>
<td>0.09</td>
<td>0.18</td>
<td>0.21</td>
<td></td>
</tr>
<tr>
<td>460</td>
<td>Leuco-amphibolite</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>50.9</td>
<td>34.00</td>
<td>3.77</td>
<td>4.91</td>
<td>1.89</td>
<td>0.37</td>
<td>0.02</td>
<td>0.05</td>
<td>0.04</td>
<td>0.05</td>
<td>0.62</td>
<td>0.66</td>
<td></td>
</tr>
<tr>
<td>461</td>
<td>Ruby mica schist</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>20.75</td>
<td>6.16</td>
<td>0.76</td>
<td>0.90</td>
<td>0.04</td>
<td>0.01</td>
<td>0.04</td>
<td>0.05</td>
<td>0.76</td>
<td>0.60</td>
<td>0.71</td>
<td></td>
</tr>
<tr>
<td>462</td>
<td>Kyanite gneiss</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>2.20</td>
<td>0.37</td>
<td>0.06</td>
<td>0.03</td>
<td>0.08</td>
<td>0.03</td>
<td>0.37</td>
<td>0.02</td>
<td>0.46</td>
<td>0.18</td>
<td>0.46</td>
<td></td>
</tr>
<tr>
<td>463</td>
<td>Ruby mica schist</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>1.76</td>
<td>0.37</td>
<td>0.04</td>
<td>0.01</td>
<td>0.05</td>
<td>0.04</td>
<td>0.37</td>
<td>0.02</td>
<td>0.44</td>
<td>0.18</td>
<td>0.44</td>
<td></td>
</tr>
<tr>
<td>464</td>
<td>Ultramafic rock</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>5.75</td>
<td>4.18</td>
<td>0.57</td>
<td>0.57</td>
<td>0.37</td>
<td>0.01</td>
<td>0.61</td>
<td>0.04</td>
<td>0.04</td>
<td>0.08</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>465</td>
<td>Ultramafic rock</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>4.18</td>
<td>3.76</td>
<td>0.57</td>
<td>0.57</td>
<td>0.37</td>
<td>0.01</td>
<td>0.61</td>
<td>0.04</td>
<td>0.04</td>
<td>0.08</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>466</td>
<td>Ultramafic rock</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>3.76</td>
<td>3.17</td>
<td>0.57</td>
<td>0.57</td>
<td>0.37</td>
<td>0.01</td>
<td>0.61</td>
<td>0.04</td>
<td>0.04</td>
<td>0.08</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>467</td>
<td>Ultramafic rock</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>3.17</td>
<td>2.88</td>
<td>0.57</td>
<td>0.57</td>
<td>0.37</td>
<td>0.01</td>
<td>0.61</td>
<td>0.04</td>
<td>0.04</td>
<td>0.08</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>11801</td>
<td>Ruby mica schist</td>
<td>65.5676</td>
<td>-52.3936</td>
<td>48.4</td>
<td>2.88</td>
<td>2.43</td>
<td>0.57</td>
<td>0.57</td>
<td>0.37</td>
<td>0.01</td>
<td>0.61</td>
<td>0.04</td>
<td>0.04</td>
<td>0.08</td>
<td>0.05</td>
<td></td>
</tr>
</tbody>
</table>

(continued on next page)
The modelled compositions of the aluminous gneiss, biotite gneiss and amphibolite are median values of the range of bulk-rock compositions reported in Szilas et al. (2016) with obvious outliers removed. The compositions used in the modelling are summarized in Table 3.

The stability of oxide and silicate minerals in high-temperature metamorphic rocks is sensitive to the amount of ferrous and ferric iron in the bulk composition (e.g., Diener and Powell, 2010; Boger et al., 2012). For the samples from Størø, the biotite-schist was assumed to have 18% of iron as ferric, 15% of the iron was assumed to be ferric for the aluminous gneiss, and 14% of the iron was assumed to be ferric for the amphibolite based on titration data from Szilas et al. (2014b) for the same lithological units in the SSB. For the ruby mica schists from Størø and Maniitsoq, various proportions of ferric to ferrous iron are examined. For the kyanite gneiss from Maniitsoq, 15% of the iron was assumed to be ferric, similar to the value used for the Størø aluminous gneiss. For the peridotite composition from Maniitsoq, 5% of the iron was assumed to be ferric, given that olivine dominated the rock and only accommodates divalent iron.

The role of silica depletion on the stability of corundum for three samples from Størø (aluminous gneiss, biotite schist and amphibolite) and one sample from Maniitsoq (kyanite gneiss) was evaluated by starting with a relatively silica-rich composition ranging to a silica-depleted composition. Two sets of diagrams were calculated to evaluate the temperature and pressure sensitivity of corundum-bearing assemblages to silica content. The first set of diagrams was constructed at a fixed pressure of 7 kbar for samples from Størø (Figs. 10a, 11a and 12a) and 10 kbar for the sample from Maniitsoq (Fig. 14a), but for a range of temperatures and SiO2 concentrations of the system. This is similar to the approach taken by Riesco et al. (2004) who evaluated silica-undersaturated rocks and corundum formation in the Susqueda Aureole in Spain. The second set of diagrams (Figs. 10b, 11b, 12b and 14b) was constructed at a fixed temperature of 800 °C at a range of pressure and SiO2 concentrations of the system for samples from Størø and Maniitsoq.

The left side of each of the temperature–composition and pressure–composition phase diagrams in Figs. 10–12 and 14 represents the silica-rich bulk composition and the right side of each diagram is the silica-depleted composition. When corundum is predicted to be stable in the phase assemblage, the amount of corundum in oxide-molar proportions (approximately equivalent to volume proportion) is illustrated by dashed contours on the appropriate diagrams.

The changes in the proportions of the different phases for two different scenarios are illustrated using mode-temperature and mode-pressure diagrams in Figs. 10–12 and in Fig. 14 for the kyanite gneiss from Maniitsoq. These diagrams represent two end-member scenarios. The first scenario assumes a constant pressure and composition, but temperature increases. This could represent the modification of the SiO2 content of the rock during interaction with an adjacent ultramafic body followed by heating during metamorphism or a scenario whereby a silica-depleted protolith undergoes prograde metamorphism (e.g., Goliani, 1989). The second scenario assumes that the system was at a constant pressure and temperature and that SiO2 was removed from the composition. This is a simplified model that could represent a rock adjacent to an ultramafic rock that has achieved thermal equilibrium but SiO2 is diffusing into the neighboring ultramafic rock (e.g., Riesco et al., 2005).

The effect of the ratio of ferric to ferrous iron in the bulk composition for the ruby-bearing rocks from Størø and Maniitsoq are explored with temperature-composition phase diagrams in Figs. 13a and 16a. The left side of these diagrams reflects only ferrous iron and the right-hand side of these diagrams represents 50% ferric iron and 50% ferrous iron.
5. Results

5.1. Bulk-rock geochemical compositions

The new geochemical data for the Maniitsoq rocks are presented in Table 1 and Fig. 7 shows selected binary plots with trends towards the ruby-mica schists. The ultramafic rocks are generally Mg-rich and very depleted, although it should be noted that several samples, which are rich in amphibole (tremolite) also have distinctly elevated SiO$_2$ of up to 55 wt.%. The kyanite gneiss is rich in SiO$_2$ and Al$_2$O$_3$ with 67 wt.% and 19 wt.%, respectively. The ruby-mica schists are characterized by having unusually high Al$_2$O$_3$ of up to 44 wt.% in combination with very low SiO$_2$ of down to 26 wt.%. There is also strong enrichment in...
components such as Th, Sr and K$_2$O for the corundum-bearing rocks (Fig. 7).

Primitive mantle-normalized trace element patterns for the Maniitsoq rocks are shown in Fig. 8. It is seen that, although the abundances vary significantly, there are common features such as troughs for Ta-Nb-Ti, as well as both positive and negative anomalies for Sr and Eu.

The geochemistry of the Storø rocks have previously been described in detail by Szilas et al. (2014b, 2016) and will thus not be repeated here. However, the Storø rocks were interpreted to have formed in a geologic environment similar to a proximal arc-related basin, because the amphibolites consistently have negative Nb-Ta-Ti anomalies (also seen in Fig. 9), and metapelites generally have a strong mafic component suggesting immature sedimentary processing. However, mature quartzites (meta-sandstone) are also present, which indicates a proximal source with a provenance from continental crust. The geochemical difference between the biotite schist and the aluminous gneiss in the SSB were interpreted by Szilas et al. (2016) to mainly be controlled by the proportion of felsic versus mafic provenance, with the aluminous gneiss being derived almost entirely from a basaltic to ultramafic source.

The main geochemical features of the median aluminous gneiss and the ruby-mica schist at the Storø locality are shown in Fig. 7. The aluminous gneiss is characterized by having relatively high SiO$_2$ and Al$_2$O$_3$ of 51 wt.% and 20 wt.%, respectively, similar to the kyanite gneiss at the Maniitsoq locality. Both the Storø aluminous gneiss and the Storø ruby-mica schist have higher TiO$_2$ than any of the Maniitsoq rocks, which is consistent with overall higher abundances of incompatible trace elements as seen in Fig. 9.

5.2. Phase equilibria models for the Storø rocks

Phase equilibria modelling of the aluminous gneiss and biotite schist compositions from Storø indicate that quartz and corundum are not stable together at equilibrium. The depletion of silica is a requirement for corundum to stabilize in the equilibrium assemblage for both compositions. For example, at 7 kbar and 800 °C, the aluminous gneiss requires a decrease of 6 wt.% SiO$_2$ to destabilize quartz and of 7 wt.% SiO$_2$ to stabilize corundum (Fig. 12a and c). The biotite schist requires a more extreme reduction of SiO$_2$ in the bulk composition to destabilize quartz (20 wt.% SiO$_2$) and stabilize corundum (22 wt.% SiO$_2$) at the same temperature and pressure (Fig. 11a and d).

The quartz-out and corundum-in phase boundaries are not sensitive to pressure, except at $P < 7$ kbar in the stability field of cordierite for the biotite schist (Fig. 11b) and at $P < 6$ kbar in the stability field of cordierite or sapphirine for the aluminous gneiss (Fig. 12b). Corundum is not predicted to be stable at $P < 6$ kbar for the aluminous gneiss due to the breakdown of corundum to produce sapphirine. The quartz-out and corundum-in field boundaries are sensitive to temperature as indicated by the shift of both to lower SiO$_2$ values at higher temperatures (Figs. 11a and 12a). This indicates that less silica depletion is required to stabilize corundum at elevated temperatures for both the aluminous gneiss and the biotite schist.

With decreasing SiO$_2$ in both the aluminous gneiss and biotite schist, corundum modes increase relatively rapidly (e.g., closer spacing of contours in Figs. 11 and 12) during the concomitant breakdown of sillimanite (Figs. 11 and 12). Sapphirine is predicted to be stable at very low SiO$_2$ concentrations (Figs. 11 and 12) and

Figure 7. Bulk-rock geochemical diagrams. Oxides are in wt.% and trace elements are in ppm. (a) CaO vs. SiO$_2$ showing a significant drop in silica content from the precursor rocks to the corundum-bearing rocks. (b) TiO$_2$ vs. MgO showing a slight increase in MgO during the corundum formation. (c) K$_2$O vs. Cr showing a strong enrichment in potassium during the corundum formation, which is consistent with the abundant biotite in this rock. (d) Al$_2$O$_3$ vs. Al$_2$O$_3$/TiO$_2$ showing a strong enrichment in alumina for the corundum-bearing rocks. (e) Th vs. Th/TiO$_2$ showing a strong enrichment in Th for the ruby mica schists. (f) Sr vs. Sr/TiO$_2$ showing a strong enrichment in Sr for the ruby mica schist. The Storø aluminous gneiss is a median composition from Szilas et al. (2016) (see Table 3) and the rest of the data are from Table 1 of this study.

Please cite this article in press as: Yakymchuk, C., Szilas, K., Corundum formation by metasomatic reactions in Archean metapelite, SW Greenland: Exploration vectors for ruby deposits within high-grade greenstone belts, Geoscience Frontiers (2017), http://dx.doi.org/10.1016/j.gsf.2017.07.008
growth of sapphirine is modelled to occur at the expense of corundum (Figs. 11c, d and 12c, d). The maximum mode of corundum occurs at SiO₂ values just above those required for sapphirine stability.

The modelled amphibolite composition does not predict corundum formation even for extreme SiO₂ depletion (Fig. 10). For this composition, silica depletion results in the disappearance of quartz at 49–47 wt.% SiO₂ over the modelled range of pressure and temperature except at T > 800 °C and P < 7 kbar, where quartz is unstable even for the modelled median composition (Fig. 10a and b). Olivine is predicted to be stable at SiO₂ < 42 wt.% for relatively low pressures at 800 °C (< 8 kbar; Fig. 10b). In general, the relatively low Al₂O₃ values of the amphibolite are expected to prohibit corundum formation due to SiO₂ depletion alone.

The predicted phase assemblage of the ruby mica schist is sensitive to the amount of ferric to ferrous iron in the bulk composition (Fig. 13a). For example, cordierite is expected to be stable in relatively oxidized rocks whereas spinel is stable in more reducing rocks (Fig. 13a). Corundum is stable over the modelled compositions but the amount of corundum is predicted to be greatest at relatively oxidizing conditions at 800 °C (Fig. 13c). The observed mineral assemblage in the ruby mica schist from Storø includes garnet, biotite, ilmenite, plagioclase, corundum and spinel. Using the P-T phase diagram in Fig. 13b, this assemblage is restricted to P > 5.5–8 with increasing temperature. This is broadly consistent with the previous estimates of the P-T conditions of metamorphism (Persson, 2007).

5.3. Phase equilibria models for the Maniitsq rocks

Phase equilibria modelling of the kyanite gneiss from Maniitsq also suggests that quartz and corundum are not stable together over the range of modelled P-T conditions, which is the same as the aluminous gneiss from Storø. At 800 °C, the first appearance of corundum is restricted to SiO₂ concentrations less than ~40 wt.% over the range of modelled pressures. This requires the extraction of ~15 wt.% SiO₂ from the starting kyanite gneiss (Fig. 14a). Note that the quartz-out and corundum-in field boundaries are not sensitive to temperature (Fig. 14a) or pressure, except at low pressures in the stability field of cordierite (P < 6 kbar; Fig. 14b). Corundum is expected to grow at the expense of kyanite or sillimanite (Fig. 14c and d).

When compared to the Storø composition, the kyanite gneiss from Maniitsq requires more SiO₂ depletion to stabilize corundum, but it can potentially produce more. For example, the Maniitsq composition is predicted to generate up to 30 mol.% corundum whereas the maximum amount of corundum in the Storø composition is ~7 mol.% due to the growth of sapphirine at the expense of corundum at low SiO₂ contents (Fig. 12a and b). No sapphirine is predicted to be stable over the modelled conditions for the Maniitsq kyanite gneiss (Fig. 14a and b).

The phase assemblage of the ruby mica schist from Maniitsq is sensitive to the amount of ferric to ferrous iron in the bulk composition (Fig. 16a). Corundum is predicted to become unstable at temperatures just below the solidus for relatively reducing compositions but is stable at the solidus for more oxidizing composition.

Figure 8. Maniitsq rock types plotted on a primitive mantle-normalized (Palme and O'Neill, 2003) element diagram. (a) Amphibolites. (b) Kyanite gneiss. (c) Ruby mica schist. (d) Ultramafic rocks. The shaded area represents the total range of the Maniitsq data (Table 1). Note the relative enrichment in Th-Zr-Hf-LREE from the kyanite gneiss to the ruby mica schists, which may suggest the involvement of melt transfer in the corundum-forming process.
Figure 9. The Storø rocks plotted on a primitive mantle-normalized (Palme and O’Neill, 2003) element diagram. Median compositions of the biotite schist and the aluminous gneiss are based on data from Szilas et al. (2016) (see Table 3). The shaded area represents the total range of the Maniitsoq data for comparison (Table 1). Note that the Storø median amphibolite is not plotted, because it is not relevant for the corundum formation at this locality.
Similar to the composition from Storø, spinel growth at the expense of corundum occurs at relatively reducing conditions (Fig. 16c).

The observed mineral assemblage in the ruby mica schist from Maniitsoq includes garnet, biotite, plagioclase, corundum and spinel. Using the P-T phase diagram in Fig. 16b, this assemblage is restricted to P > 7.5–10 kbar with increasing temperature from 650 to 800 °C. Sapphirine is predicted to be stable at lower pressures, but this mineral was not observed in the Maniitsoq sample.

The peridotite from Maniitsoq is predicted to contain a stable assemblage of plagioclase, olivine, spinel and orthopyroxene (Fig. 15) over the range P-T conditions predicted for the ruby mica schist (Fig. 13). The absence of garnet from the peridotite restricts the pressure to <14–15 kbar with increasing temperature from 700 to 900 °C, which is compatible with the P-T estimates of Dyck et al. (2015).

### 6. Discussion

6.1. Mechanism for corundum stabilization

Corundum-bearing rocks in metamorphic terranes can be produced by (1) hydrothermal alteration (e.g., Bottrill, 1998), (2) metasomatic exchange with the ultramafic rocks or rodingitization (e.g., Riesco et al., 2005) and (3) melt loss (e.g., Cartwright and Barnicoat, 1986). A corollary of these processes is that the chemistries of the nearby amphibolites at Maniitsoq are not appropriate for determining the geotectonic setting of formation. In the trace element diagram in Fig. 8a, some samples clearly have crustal signatures with elevated La, Pb and Th relative to Nb and Ta, although two samples have flatter patterns. Thus, we will not make inferences about the geodynamic environment of formation based on these data, but will leave that discussion for a future study targeting Maniitsoq mafic rocks on a regional basis.

The role of hydrothermal alteration in the genesis of the corundum-bearing rocks can be assessed using the major and trace element chemistries. In Fig. 7 we show selected bulk-rock geochemical plot for the Storø and Maniitsoq rocks. Potassium and Sr are strongly enriched in the corundum-bearing rocks, which is hardly surprising due to their relatively volatile behavior during fluid-movement. It should be noted that TiO₂ is not generally much different between the metapelite precursor rocks and the corundum-bearing metasomatic products. Aluminum oxide versus TiO₂ is a ratio that is usually considered constant during hydrothermal alteration processes, because both elements have high field-strength (HFS) and are not easily mobilized. However, we observe that although TiO₂ is not significantly different for these rocks, they show a large difference in their Al₂O₃ contents and the corundum-bearing rocks even show a slight increase in this ratio despite TiO₂ actually being marginally higher in these. This suggests that Al₂O₃ was mobilized in these rocks, which either points to melt mobilization, alkaline hydroxide-complexation, or mobilization by alkali-bearing high-temperature fluids as Na-Al-Si-O polymers (e.g., Kerrick, 1988; Oelkers et al., 1994; Manning, 2006). Thorium is also a HFS element, which is not typically mobilized during hydrothermal processes. However, Kessel et al.

### Table 2

Phase abbreviations used in the figures of the present study (from Holland and Powell, 2011).

<table>
<thead>
<tr>
<th>Phase</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>augite</td>
<td>aug</td>
</tr>
<tr>
<td>biotite</td>
<td>bi</td>
</tr>
<tr>
<td>cordierite</td>
<td>cd</td>
</tr>
<tr>
<td>corundum</td>
<td>cor</td>
</tr>
<tr>
<td>garnet</td>
<td>g</td>
</tr>
<tr>
<td>H₂O</td>
<td>H₂O</td>
</tr>
<tr>
<td>hornblende</td>
<td>hb</td>
</tr>
<tr>
<td>ilmenite</td>
<td>il</td>
</tr>
<tr>
<td>K-feldspar</td>
<td>ksp</td>
</tr>
<tr>
<td>kyanite</td>
<td>ky</td>
</tr>
</tbody>
</table>

### Table 3

Compositions used in phase equilibria modelling including the new bulk-rock geochemical data for the Maniitsoq rocks, as well as existing data for the Storø rocks from Szilas et al. (2016).

<table>
<thead>
<tr>
<th>Store area</th>
<th>Figure</th>
<th>mol%</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>SiO₂</td>
<td>TiO₂</td>
</tr>
<tr>
<td>Storø area</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Amphibolite</td>
<td>x = 9</td>
<td>3.20</td>
</tr>
<tr>
<td>x = 11</td>
<td></td>
<td>3.92</td>
</tr>
<tr>
<td>Aluminous gneiss</td>
<td>x = 11</td>
<td>1.01</td>
</tr>
<tr>
<td>Biotite schist</td>
<td>x = 11</td>
<td>1.41</td>
</tr>
<tr>
<td>Ruby mica schist</td>
<td>x = 10</td>
<td>2.86</td>
</tr>
<tr>
<td>x = 11</td>
<td></td>
<td>5.43</td>
</tr>
<tr>
<td>Ruby mica schist</td>
<td>x = 12a</td>
<td>34.31</td>
</tr>
<tr>
<td>x = 12b</td>
<td></td>
<td>33.30</td>
</tr>
<tr>
<td>P–T</td>
<td></td>
<td>34.25</td>
</tr>
<tr>
<td>Maniitsoq area</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Kyanite gneiss</td>
<td>x = 13</td>
<td>0.56</td>
</tr>
<tr>
<td>x = 13</td>
<td></td>
<td>1.61</td>
</tr>
<tr>
<td>Ruby mica schist</td>
<td>x = 15a</td>
<td>41.47</td>
</tr>
<tr>
<td>x = 15b</td>
<td></td>
<td>40.31</td>
</tr>
<tr>
<td>P–T</td>
<td></td>
<td>41.29</td>
</tr>
</tbody>
</table>

1° in excess, ″ not considered.
Figure 10. Storø amphibolite (for median composition see Table 3). (a) Temperature-$\text{SiO}_2$ diagram calculated for a fixed pressure of 7 kbar. (b) Pressure-$\text{SiO}_2$ diagram for the median amphibolite composition calculated for a fixed temperature of 800°C. The median $\text{SiO}_2$ content is shown by the arrow on the horizontal axis. No corundum is expected to be stable in this composition across the modelled compositional range.


(2005) demonstrated that Th can indeed become mobilized in high-temperature fluids or in a melt phase, as it is an incompatible trace element, which is also observed for major rock-forming elements. Nonetheless, hydrothermal alteration is not the dominant process for generating corundum-bearing rocks at Storø and Maniitsoq. An attempt was made to apply the isoclin method of Grant (1986) to make mass-balance calculations for the metasomatic transformation from the metapelites at Store and Maniitsoq to the protoliths of the corundum-bearing schists. However, this approach was not possible due to significant differences in the ratios of refractory elements between the metapelitic rocks and the resulting corundum schists. Thus, despite a close spatial proximity and a general similarity in trace element patterns between the sample pairs, the systematics of their high-grade metapelites at Storø and Maniitsoq to the protolith suitable for corundum formation at higher metamorphic grade. This observation is compatible with three different explanations: either (1) partial melting was involved, which would fractionate fluid-immobile trace elements, and/or (2) mechanical mixing of pelite and ultramafic rocks during their juxtaposition resulted in erratic trace element ratios and differences to the actual protolith of the corundum-bearing schists making the isoclin method inapplicable. (3) Alternatively, and perhaps the simplest explanation is that the metapelites were heterogeneous at a scale smaller than the sampling and thus any primary mineralogical heterogeneity would also control the budget of immobile elements. Nonetheless, hydrothermal alteration is not the dominant process for generating corundum-bearing rocks at Storø and Maniitsoq.

Metasomatic exchange or diffusion of silica between silica-rich and silica-poor rocks during the metamorphic grade.

Figure 11. Store biotite schist (for median composition see Table 3). (a) Temperature-SiO2 diagram calculated for a fixed pressure of 7 kbar. (b) Pressure-SiO2 diagram for the median biotite schist composition calculated for a fixed temperature of 800 °C. The median SiO2 content is shown by the arrow on the horizontal axis. Contours represent the mode of the corundum (~vol.3). Quartz and corundum are never stable together. (c) Modebox diagram calculated at 46.4 wt.% SiO2 for the Storø biotite schist composition over a range of temperatures. (d) Isothermal modebox diagram at 800 °C for various SiO2 compositions of the biotite schist. The oxide-molar proportions are approximately equivalent to volumetric proportions (e.g., modes).

Please cite this article in press as: Yakymchuk, C., Szilas, K., Corundum formation by metasomatic reactions in Archean metapelite, SW Greenland: Exploration vectors for ruby deposits within high-grade greenstone belts, Geoscience Frontiers (2017), http://dx.doi.org/10.1016/j.gsf.2017.07.008
rocks in high-grade gneiss terranes (e.g., Rieko et al., 2004, 2005). The driving force for diffusion in metamorphic and metasomatic systems is differences in chemical potential gradients of constituent components (e.g., White et al., 2008). For the ruby-bearing rocks, the difference in chemical potential of silica between the relatively silica-poor ultramafic rock and the adjacent silica-rich aluminous rocks is expected to be the principal factor. The juxtaposition of these rocks would result in a chemical potential gradient that could be equalized by the transfer of silica from the rock with the higher chemical potential (aluminous gneiss) to the rock with lower silica content (ultramafic rock).

THERMOCALC. For Storø, the calculated chemical potential of SiO2 ($\mu_{SiO2}$) at P–T conditions of 800 °C and 7 kbar of the ultramafic composition is −981.76 kJ/mol whereas it is a much higher value of −973.65 kJ/mol for the aluminous gneiss. The calculated value for the ruby-mica schist is −976.37 kJ/mol, which is in between the other two values. For Maniitsoq, the $\mu_{SiO2}$ value at P–T conditions of 800 °C and 10 kbar for the ultramafic composition is −975.95 kJ/mol. The kyanite-gneiss has a value of −966.64 kJ/mol and the ruby-bearing rock has an intermediate value of −972.57 kJ/mol. Both examples show that there is a strong gradient in chemical potential that develops at or near the peak of metamorphism between the ultramafic rock and the adjacent aluminous rocks. The ruby-bearing rocks are interpreted to represent a snapshot of the system attempting to equilibrate this gradient by diffusion of silica from the relatively high-$\mu_{SiO2}$ aluminous rocks towards the relatively low-$\mu_{SiO2}$ ultramafic rocks. This is supported by the estimated intermediate $\mu_{SiO2}$ values of the ruby-bearing rocks compared with the ultramafic rocks and aluminous gneisses.
Melt loss is an additional mechanism that may form corundum-bearing assemblages. The composition and proportion of melt for the Storø aluminous gneiss and Maniitsoq kyanite gneiss were modeled in THERMOCALC and the results are summarized in Fig. 17. The composition of the melt are those in equilibrium with both rock types at 700, 750, 800, 850 and 900 °C and the melt loss vectors represent the residuum composition after the extraction of 10 mol.% (roughly equivalent to 10 vol.%) melt from the system. This amount is considered a maximum because the calculated proportion of melt at the modeled P–T conditions is generally <10 mol.% (Figs. 12c, d and 14c, d). Vectors representing the composition of the residue after melt loss plot towards the FeO + MgO apex and away from the SiO2 apex in Fig. 17 resulting in slightly more residual compositions but not enough to account for the measured compositions of the corundum-bearing rocks. Nonetheless, melt loss may have made a minor contribution to reducing the silica content of the inferred protoliths.

The relative importance of melt loss and silica depletion during metasomatism and corundum formation is evaluated in Fig. 17. For the Maniitsoq sample, some combination of melt loss and silica depletion is plausible to explain the composition of the ruby-bearing rocks. The modelled compositions do not produce the measured compositions exactly, but it is the correct trajectory and within the range of the total variability of corundum-bearing mica schists at Maniitsoq. For the Storø aluminous gneiss, the melt loss vector is incompatible with the observed trends, and it should also be noted that the P–T conditions for this sample do not support extensive partial melting. However, as seen in Fig. 9, the Storø ruby mica schist also appears to be enriched in Th-Zr-Hf-LREE relative to the aluminous gneiss precursor rock, which does indicate a role for melt transfer or alternatively a high-temperature supercritical fluid in which these elements behaved strongly incompatible. The ruby-mica schists may require prior Al2O3 enrichment (or FeO + MgO depletion) via hydrothermal leaching, however, the very large Al-enrichment observed in Fig. 7 requires significant mobilization either as Al-hydroxide complexes or as Na-Al-Si-O polymers (Oelkers et al., 1994; Manning, 2006).

It is important to note the large enrichment in silica in Fig. 17, from the most refractory peridotite (sample 455) to the most amphibole-rich peridotite (sample 466) at the Maniitsoq corundum locality. This clearly demonstrates that the buffering capacity of the ultramafic rocks with respect to silica is of great importance.
because it is of the same magnitude as would be required for the depletion of the aluminous precursor to the composition of the corundum orthometa- plite. We consider this observation to be strong evidence for the direct involvement of the ultrama- sons. Support for this model is seen by the extensive frication that would in the mobility of primarily silica to form tremolite in the ultrama- rocks. Alternatively, the silica transfer could have occurred during the formation of tremolite in the ultramafic rocks at higher temperature conditions. This would result in a significant volume decrease of the ultramafic rocks and fracturing of the surrounding metapelitic to accommodate this volume change, which could induce fluid infiltration that would result in the mobility of primarily silica to form tremolite in the ultramafic rocks. Support for this model is seen by the extensive network of amphibole dominated fractures close to the contact between the ultramafic rocks and the kyanite gneiss (Fig. 5c).

Nevertheless, the loss of silica by itself cannot account for the strong enrichment in Al$_2$O$_3$ that is needed to stabilize corundum to the extent that is seen in the sample of the Storø ruby schist, where Al$_2$O$_3$ comprises up to 30 mol.% of the measured composition. Two
ways of doing this are by either: (1) pre-metamorphic hydrothermal leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

dom leaching of the aluminous gneiss, which would serve to cause residual enrichment in fluid-immobile components, or (2) by hydrothermal addition of Al-hydroxide complexes under highly alkaline conditions. The former is not supported because TiO₂, which would certainly have remained immobile, is generally within the same range between the metapelites and the corundum-bearing rocks. The second option of introduction of Al-hydroxide complexes during hydrothermal alteration is rather speculative, but may be supported by the reducing and alkaline nature of serpentinitization reactions (e.g., Frost, 1985; Klein et al., 2013). This is perhaps supported by the abundant calc-silicate minerals within the leucocamphibolites at Maniitsoq, such as epidote, diopside and garnet, which is typical of rodingitization, a process which requires coamphibolites at Maniitsoq, such as epidote, diopside and garnet, supported by the abundant calc-silicate minerals within the leu-

6.2. Limitations of modelling

The phase equilibria modelling presented here suggests that quartz and corundum do not form a stable assemblage together in relatively high-temperature metamorphic rocks. This contrasts with some reports of assemblages that include quartz + sillimanite + corundum (e.g., Guiraud et al., 1996; Shaw and Arima, 1998). Experimental work by Harlov et al. (2008) demonstrated that these assemblages are metastable, but may persist in metamorphic rocks due to sluggish kinetics. However, quartz in direct contact with corundum is relatively rare in natural samples and corundum + quartz assemblages are usually found in granulites with complex microstructures (e.g., White et al., 2008; Kelsey and Hand, 2015) that may not reflect hand-sample scale equilibrium, which is what we model here.

Phase equilibria modelling is a simplification of natural systems. Here, we investigate two occurrences of ruby from Store and Maniitsoq using median and measured compositions of key units to evaluate the general framework of corundum and quartz stability during metamorphism. Therefore, the results here are not necessarily representative of all bulk compositions that can host ruby. For example, modelling of the Storø aluminous gneiss compositions predicts a few volume percent corundum in the rocks. However, there are occurrences in the Storø region where corundum modes can locally reach 10%–20% (Fig. 4). Nonetheless, the general reactions and phase assemblages modelled provide some important general criteria for ruby exploration in metamorphic rocks.

We focus on SiO₂ depletion and, to a lesser extent, the ratio of ferric to ferrous iron in the bulk composition to explore the stability of corundum. However, it is unlikely that SiO₂ is the only mobile component in metasomatic systems that contain ruby and the geochemistry of the rocks from Storø and Maniitsoq may indicate Al₂O₃ mobility. Riesco et al. (2004) demonstrated that SiO₂ and Na₂O depletion affect the stability of corundum in metapelites from the Eastern Alps. Furthermore, we did not investigate the role of fluid composition on the subsolidus stability of corundum-bearing assemblages, but CO₂-rich fluids may be important in some ruby occurrences in graphite-rich host rocks, such as those found in southeast Kenya (e.g., Mercier et al., 1999). Schumacher et al. (2011) suggested that high pH at low Eh might have played a role in the formation of the ruby in the Fiskenæsset complex from alkaline fluids (Keulen et al., 2014).

6.3. Implications for mineral exploration

There are five main implications of the phase equilibria modeling for ruby exploration in metamorphic terrains. Firstly, if quartz is present then corundum is not expected to be stable. Quartz-bearing rocks should not be prioritized for ruby exploration. Secondly, less SiO₂ depletion of the bulk composition is required at higher temperatures to stabilize corundum (Figs. 11a and 12a). Therefore, in general, higher-grade metamorphic terrains are potentially better targets for ruby exploration than low grade ones. However, when spinel and sapphireine are stabilized at higher temperatures, they grow at the expense of corundum. The association of sapphireine and ruby has been documented in deposits at Qeqertarsuatsiaat (Fiske-næsset) in SW Greenland (e.g., Herd et al., 1969) and in southeast Kenya (Mercier et al., 1999). Thirdly, the main corundum producing reaction includes sillimanite or kyanite as a reactant. Sillimanite (or kyanite) bearing rocks should be targeted because they may have enough Al₂O₃ to stabilize corundum if some SiO₂ was locally mobilized out of them. Fourthly, aluminous protoliths should be targeted because more mafic rocks will not produce corundum, even with extreme SiO₂ depletion (Fig. 10). Finally, a geological mechanism to deplete the aluminous protolith in SiO₂ is required. At the Storø and Maniitsoq ruby occurrences, this is accomplished by the juxtaposition of low silica ultramafic rocks with aluminous gneisses. The juxtaposition of ultramafic rocks and aluminous rocks has also been proposed as a model for ruby-formation in the Fiskenæsset complex (Schumacher et al., 2011; Keulen et al., 2014).

Ultramafic rocks are relatively common in the region between Nuuk and Maniitsoq (e.g., Szilas et al., 2015), and as documented in the present study (see Section 3), such rocks may indeed become teconically juxtaposed with aluminous supracrustal rocks and result in the formation of corundum during subsequent high-grade
Figure 16. Maniitsoq ruby mica schist (sample 464). (a) Temperature-composition diagram for various ratios of ferric to ferrous iron for the ruby mica schist from Maniitsoq calculated at a fixed pressure of 10 kbar. (b) Pressure-temperature phase diagram for the ruby mica schist from Maniitsoq. (c) Composition-mode diagram illustrating the slight increase in the mode of corundum at the expense of spinel towards more oxidized compositions. (d) Temperature-mode diagram showing the decrease in the predicted mode of corundum towards higher temperatures due to the growth of spinel and sapphire. Note that this particular sample is relatively silica-rich and has the lowest Al₂O₃ content of any of the studied corundum-bearing rocks. This is clearly reflected by the low corundum mode and the sample likely marks the lower boundary for composition that stabilize corundum. However, the adjacent ruby mica sample (463) has the highest Al₂O₃ and lowest SiO₂ of any of the studied rocks, and would thus likely have a corundum mode larger than even Storø sample 11801 of 20%. 

Please cite this article in press as: Yakymchuk, C., Szilas, K., Corundum formation by metasomatic reactions in Archean metapelite, SW Greenland: Exploration vectors for ruby deposits within high-grade greenstone belts, Geoscience Frontiers (2017), http://dx.doi.org/10.1016/j.gsf.2017.07.008
metamorphism. The association between ultramafic rocks and corundum in aluminous supracrustal rocks in metamorphic terrains has also been observed in Proterozoic and Phanerozoic metamorphic terrains (Mercier et al., 1999; Riesco et al., 2005).

We recommend that future work on the Storø and Maniitsoq corundum localities should investigate the zoning patterns of the euhedral rubies and examine the occurrence of sapphire vs. sapphirine in these rocks, which are easily mistaken in the field. Detailed in situ trace element and oxygen isotope measurements are also recommended to expand the current data base for corundum from Greenland.

7. Conclusions

Ruby-bearing rocks at Storø and Maniitsoq are found at the contact between ultramafic rocks and aluminous gneisses. Trace element chemical composition of the ruby-bearing rocks indicate that their protoliths were similar to the adjacent aluminous gneisses, which are inferred to have a sedimentary origin. Tectonic juxtaposition of these rock types and metamorphism created a chemical potential gradient in silica that drove desilicification of the aluminous gneisses and silicification of the ultramafic rocks. The ultramafic rocks also appear to be important as a source of trace
elements needed for the coloring of corundum (ruby/sapphire), such as chromium and titanium. Phase equilibria modeling of the aluminous gneisses indicates that silica depletion during metamorphism can destabilize quartz and grow corundum at both localities at the estimated P–T conditions. However, the modeling also predicts that too high temperature (>750 °C) promotes the stability of sapphire at the expense of corundum, and too low pressure (<6 kbar) stabilizes cordierite. Changes in iron speciation (Fe<sup>3+</sup>/Fe<sup>2+</sup>) have only a minor effect on the predicted mode of corundum at the estimated peak metamorphic conditions. Aluminum appears to have been enriched in the corundum-bearing rocks relative to their metasedimentary precursor rocks. Melt extraction cannot reproduce the measured compositions but Al-hydroxide complexes or Na-Al-Si-O polymers may have transported Al by metasomatic reactions in the deep crust. An overall high prospectivity of corundum in Archean greenstone belts is thus found in amphibole-facies terrains hosting sillimanite- or kyanite-rich aluminous gneisses, which are spatially related to metaperidotites as demonstrated by ruby-bearing rocks at Store and Manitosq.

Acknowledgements

The Ministry of Mineral Resources (MMR) of the Greenland Government is acknowledged for supporting the field and analytical work in the Manitosq region, SW Greenland, where the main samples of this study were collected. K. Sizlas would like to thank the CARLSBERG FOUNDATION for support to carry out this work via grant CF16-0059. We appreciate comments on an early version of the manuscript by Julie A. Hollis, Anette Juul-Nielsen, Nynke T. Keulen and Agnete Steenfelt. Constructive critiques by Hugh Rollinson and an anonymous reviewer improved the resulting paper.

References


