Basic characteristics and kinetics of degradation in aqueous buffer of selected diclofenac prodrugs intended for joint injection
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Basic characteristics and kinetics of degradation in aqueous buffer of selected diclofenac prodrugs intended for joint injection

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Introduction

Recently, a novel ester prodrug approach for the accomplishment of local and sustained diclofenac action after injection into joints was reported (1). It is to be expected that both onset and duration of diclofenac action can be modified by variation of inherent ester prodrug properties including their pH dependent solubility and charge as well as their susceptibility to undergo esterase facilitated hydrolysis. In the present study, three-diclofenac ester prodrugs differing with respect to the spacer carbon chain length (Fig. 1) were synthesized and evaluated in vitro. Thus, the objectives of the present study were:

(i) to determine the effect of the spacer chain length on the pK_a value and aqueous pH-dependent solubility of the prodrugs.
(ii) to investigate the kinetics and mechanism of degradation of the three prodrugs in aqueous solution in the pH-range 1-10 as well as in 80 % (v/v) human synovial fluid (SF) and 80 % (v/v) plasma at 37°C.
(iii) to characterize in vitro release of diclofenac from prodrug suspensions using the rotating dialysis cell model.

Mechanism of degradation

The stability of the diclofenac ester prodrugs after incubation in human 80 % (v/v) SF and 80 % (v/v) plasma was studied at 37°C. Compared to the stability in 67 mM phosphate buffer solution at pH 7.4, the prodrugs underwent much faster degradation in the biological media indicating the involvement of enzyme-mediated prodrug conversion (Table 1) to yield the active diclofenac.

Stability in human plasma and synovial fluid

The stability of the diclofenac ester prodrugs after incubation in human 80 % (v/v) SF and 80 % (v/v) plasma was studied at 37°C. Compared to the stability in 67 mM phosphate buffer solution at pH 7.4, the prodrugs underwent much faster degradation in the biological media indicating the involvement of enzyme-mediated prodrug conversion (Table 1) to yield the active diclofenac.

Table 1: pK_a, half-lives and solubility (mean ± S.D., n=3) of prodrugs in aqueous solution at pH 7.4, 80 % (v/v) SF and 80 % (v/v) plasma at 37°C.

<table>
<thead>
<tr>
<th>Prodrug</th>
<th>pH 7.4</th>
<th>pH SF</th>
<th>pH Plasma</th>
</tr>
</thead>
<tbody>
<tr>
<td>DP-1</td>
<td>7.9 ± 0.3</td>
<td>6.4 ± 0.6</td>
<td>6.2 ± 0.4</td>
</tr>
<tr>
<td>DP-2</td>
<td>7.5 ± 0.3</td>
<td>6.5 ± 0.5</td>
<td>6.2 ± 0.5</td>
</tr>
<tr>
<td>DP-3</td>
<td>7.5 ± 0.3</td>
<td>6.5 ± 0.5</td>
<td>6.2 ± 0.5</td>
</tr>
</tbody>
</table>

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